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#### Key indicators

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.009 Å Disorder in main residue R factor = 0.053 wR factor = 0.152 Data-to-parameter ratio = 11.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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# 1-(4-Bromo-2-fluorophenyl)-3-trifluoromethyl-1*H*-pyrazol-5-yl benzenesulfonate

The molecular structure of the title compound,  $C_{16}H_9BrF_4$ - $N_2O_3S$ , is stabilized by an intramolecular  $C-H\cdots O$  hydrogen bond. In the crystal structure, supramolecular layers are formed parallel to the *bc* plane by an intermolecular  $C-H\cdots F$  hydrogen bond. The  $CF_3$  group was found to be disordered.

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### Comment

3-(Trifluoromethyl)pyrazole derivatives have active fungicidal properties (Hwang *et al.*, 1994; Liu & Li, 2004), as well as insecticidal activities (Kim *et al.*, 1989). In a search for new compounds with improved properties, the title compound, (I), was obtained *via* benzenesulfonation of 1-(4-bromo-2-fluorophenyl)-3-trifluoromethyl-1*H*-pyrazol-5-one. The crystal structures of some related compounds have been reported previously (Li, Duan *et al.*, 2005; Li, Li *et al.*, 2005).



The molecular structure of (I) is illustrated in Fig. 1 and shows that the rings are all planar. The dihedral angles between the pyrazole and the benzene rings at N1 and S1 are 60.1 (3) and 50.9 (3)°, respectively.

The molecular structure of (I) is stabilized by an intramolecular C—H···O hydrogen bond (Table 1). In the crystal structure (Fig. 2), supramolecular layers are formed parallel to the *bc* plane by an intermolecular C16—H16···F1 hydrogen bond. There are no significant interactions between these layers.

### **Experimental**

Benzenesulfonyl chloride (0.35 g, 2.0 mmol) in benzene (6 ml) was added dropwise to a suspension of 1-(4-bromo-2-fluorophenyl)-3-trifluoromethyl-1*H*-pyrazol-5-one (0.65 g, 2.0 mmol) [prepared according to the literature method of Liu & Li (2004)], anhydrous

sodium carbonate (0.21 g, 2.0 mmol), a catalytic amount of tetrabutylammonium chloride in benzene (10 ml) and water (1 ml), over approximately 30 min at 283 K. The resulting mixture was stirred at room temperature for an additional 1 h. The benzene layer was collected and evaporated under reduced pressure. The crude product was recrystallized from ethyl acetate/petroleum ether (1:1  $\nu/\nu$ ) to give (I) as a colorless solid (0.80 g, yield: 86%; m.p. 342–343 K). Suitable single crystals were grown from a solution in ethyl acetate/*n*-hexane (1:1  $\nu/\nu$ ).

> $D_x = 1.724 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 1907 reflections  $\theta = 2.4-22.6^{\circ}$  $\mu = 2.47 \text{ mm}^{-1}$

T = 293 (2) K

 $\begin{aligned} R_{\text{int}} &= 0.058\\ \theta_{\text{max}} &= 25.0^{\circ}\\ h &= -23 \rightarrow 23\\ k &= -11 \rightarrow 11\\ l &= -11 \rightarrow 8 \end{aligned}$ 

Block, colorless

 $0.30 \times 0.24 \times 0.20 \text{ mm}$ 

3161 independent reflections 1656 reflections with  $I > 2\sigma(I)$ 

#### Crystal data

$C_{16}H_9BrF_4N_2O_3S$
$M_r = 465.22$
Monoclinic, $P2_1/c$
a = 19.740 (4)  Å
b = 9.5361 (18) Å
c = 9.7230 (18)  Å
$\beta = 101.699 \ (3)^{\circ}$
V = 1792.3 (6) Å <sup>3</sup>
Z = 4

#### Data collection

Bruker SMART CCD area-detector	
diffractometer	
$\varphi$ and $\omega$ scans	
Absorption correction: multi-scan	
(SADABS; Sheldrick, 1996)	
$T_{\min} = 0.525, \ T_{\max} = 0.638$	
8850 measured reflections	

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.0602P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.053$	+ 2.3366P]
$wR(F^2) = 0.152$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.01	$(\Delta/\sigma)_{\rm max} = 0.001$
3161 reflections	$\Delta \rho_{\rm max} = 0.84 \text{ e} \text{ Å}^{-3}$
272 parameters	$\Delta \rho_{\rm min} = -0.71 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} C12-H12\cdots O2\\ C16-H16\cdots F1^i \end{array}$	0.93	2.56	2.927 (10)	104
	0.93	2.52	3.428 (7)	166

Symmetry code: (i)  $x, -y + \frac{3}{2}, z + \frac{1}{2}$ 

H atoms were positioned geometrically (C-H = 0.93 Å) and constrained to ride on their parent atoms, with  $U_{iso}(H) = 1.2U_{eq}(C)$ . The CF<sub>3</sub> group was found to be disordered and two distinct conformations were observed. The site occupancies refined to 0.623 (2) and 0.377 (2).

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.



Figure 1

The molecular structure of compound (I), showing the atom-numbering scheme and displacement ellipsoids drawn at the 30% probability level. The dashed line indicates an intramolecular hydrogen bond. Only the major component of the disordered  $CF_3$  group is shown.



#### Figure 2

A crystal packing diagram for compound (I).  $C-H \cdots F$  hydrogen bonds are indicated by dashed lines. Only the major component of the disordered CF<sub>3</sub> group is shown.

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